This article was downloaded by: On: 23 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37- 41 Mortimer Street, London W1T 3JH, UK



## Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information: <http://www.informaworld.com/smpp/title~content=t713617200>

## A NEW APPROACH TO A DISACCHARIDIC HAPTEN CONTAINING A GALACTOFURANOSYL ENTITY1-2[1], [2]]

Vincent Ferrières<sup>a</sup>; Myriam Roussel<sup>a</sup>; Muriel Gelin<sup>a</sup>; Daniel Plusquellec<sup>a</sup> a Ecole Nationale Supérieure de Chimie de Rennes, Synthèses et Activations de Biomolécules, CNRS UMR 6052, Institut de Chimie de Rennes, Rennes, France

Online publication date: 02 July 2002

To cite this Article Ferrières, Vincent , Roussel, Myriam , Gelin, Muriel and Plusquellec, Daniel(2001) 'A NEW APPROACH TO A DISACCHARIDIC HAPTEN CONTAINING A GALACTOFURANOSYL ENTITY1-2[1], [2]]', Journal of Carbohydrate Chemistry, 20: 9, 855 — 865

To link to this Article: DOI: 10.1081/CAR-100108662 URL: <http://dx.doi.org/10.1081/CAR-100108662>

# PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use:<http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

#### J. CARBOHYDRATE CHEMISTRY, 20(9), 855–865 (2001)

## **A NEW APPROACH TO A DISACCHARIDIC HAPTEN CONTAINING A GALACTOFURANOSYL ENTITY1, 2**

## **Vincent Ferrières,\* Myriam Roussel, Muriel Gelin, and Daniel Plusquellec**

Ecole Nationale Supérieure de Chimie de Rennes, Synthèses et Activations de Biomolécules, CNRS UMR 6052, Institut de Chimie de Rennes, Avenue du Général Leclerc, F-35700 Rennes, France

### **ABSTRACT**

A short synthetic entry into the disaccharidic hapten  $\beta$ -D-Galf-(1→3)- $\alpha$ -D- $\text{Map-O}(\text{CH}_2)_8\text{CO}_2\text{Me}$  containing a galactofuranosyl entity at the non-reducing part is described. The synthetic scheme was designed in such a way that each required building block could be obtained by minimizing the number of chemical and purification steps. Indeed, compound **8** was obtained according to a four step–one pot preparation.

#### **INTRODUCTION**

Worldwide each year, a large number of illnesses due to parasites are reported. This can be explained by the natural outstanding diversity of parasites which carry pathogenic agents and by their high potential of adaptability to the contaminated host.<sup>3</sup> Generally, host and parasite are able to live in harmony. Nevertheless, *Aspergillus*, *Leishmania* and *Trypanosoma* species induce severe diseases. For instance, *T. gambiense* and *T. rhodesiense* are responsible for sleeping sickness in West and East Africa, respectively, while American or Chagas disease originates from *T. cruzi*. <sup>3</sup> Moreover, increasing human migrations result in exotic

<sup>\*</sup>Corresponding author.





*Scheme 1.* Retrosynthetic scheme for the preparation of disaccharide 1.

parasitoses from African and South American continents appearing more frequently in the northern hemisphere. The clinical effects of these diseases may be caused by a number of biomolecules such as proteins, glycoproteins and/or glycolipids. Among the latter, it was established that the epitopic unit of glycoconjugates, carried by infective microorganisms, contains galacto*furanose* (Gal*f*) residues as the non-reducing end.<sup>4</sup>

Considering our interest in hexo*furanose* chemistry,<sup>5</sup> we present herein a convenient access to the disaccharidic hapten  $1$  (Scheme 1) characterized by a  $\beta$ -D-Gal*f* entity and a functionalized spacer for further connection to a carrier. Our approach relies on (i) a specific synthesis of galactofuranosyl donors **2**, (ii) a concise  $\alpha$ -D-mannosylation of the spacer arm *via* the persilylated mannopyranosyl iodide **4** and (iii) a highly regioselective furanosylation reaction of diol **3**.

## **RESULTS AND DISCUSSION**

The known octyl galactofuranoside **5** was obtained from D-Gal according to a procedure developed in our laboratory<sup>6</sup> and then benzoylated (Scheme 2). Subsequent acetolysis under conditions worked out for the preparation of per-*O*acetyl-hexofuranoses<sup>7</sup> was unfortunately unsuccessful, probably because of greater electron withdrawing effects of benzoyl groups as compared to those of acetyl units. Therefore, substituting acetic anhydride by the more reactive trifluoroacetic anhydride under carefully controlled acidic conditions afforded the anomeric trifluoroacetyl furanose **7** without ring expansion. This intermediate



Copyright © Marcel Dekker, Inc. All rights reserved

MARCEL DEKKER, INC. 270 Madison Avenue, New York, New York 10016





*Scheme 2.* Preparation of galactofuranosyl donors **2a** and **2b**.

could be isolated as a mixture of  $\alpha$ ,  $\beta$  -anomers ( $\alpha/\beta$  = 1:4.6) but was best used for further thioglycosidation just after simple work-up. Reaction of **7** with thiophenol or ethanethiol, promoted by boron trifluoride-diethyl ether complex, resulted in the specific formation of  $\beta$ -thioglycofuranosides  $2a$  and  $2b$ , respectively.

Parallel to this sequence, we also focused our attention on a short entry into the mannopyranosidic acceptor **3**. Amongst various hypotheses, we considered the glycosyl iodide approach<sup>8</sup> in order to shorten the time consuming preparation of  $3$ . Indeed, persilylation of free D-mannose (D-Man) occurred in pyridine in 30 min using hexamethyldisilazane (HMDS) and chlorotrimethylsilane (TMSCl) as silylating agents (Scheme 3). The resulting anomeric mixture ( $\alpha/\beta = 1:6.7$ ), exclusively in the pyranose form as revealed by NMR analysis, quantitatively reacted with iodotrimethylsilane (TMSI) to give the key donor **4**. Glycosylation of 8-ethoxycarbonyloctanol was then carried out in dichloromethane in the presence of sterically hindered 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP) used as acid scavenger. The glycosylation step was followed by simple *in situ* methanolysis of silyl ethers. However, under these conditions, a first attempt to mannosylate the  $\omega$ -functionalized alcohol yielded an anomeric mixture of mannopyranoside  $\mathbf{8}\alpha, \beta$ . To overcome this problem, acetylation of  $\mathbf{8}\alpha$ ,  $\beta$  followed by anomerization under the action of ferric chloride<sup>10</sup> and transesterification afforded the required pure  $\delta \alpha$ . Nevertheless, in order to shorten this multi-step sequence, we expected that a diastereospecific mannosidation would occur using the well known *in situ* anomerization approach.<sup>11</sup> Therefore, after optimization, best results were obtained by: (i) synthesizing the donor **4** from persilylated mannopyranose in 45 min, (ii) effecting the anomerization catalyzed by tetrabutylammonium iodide (TBAI) in 15 min, and



**Reagents and conditions:** (a) HMDS, TMSCl, Py; (b) TMSI, CH<sub>2</sub>Cl<sub>2</sub>; (c) TBAI,  $HO(CH<sub>2</sub>)<sub>8</sub>CO<sub>2</sub>Et$ , DTBMP; MeOH (55% from D-Man); (d) PhCH(OMe)<sub>2</sub>, ZnCl<sub>2</sub>, AcOEt (44%).

*Scheme 3.* Synthesis of the mannopyranosidic building block **3**.



#### **858 FERRIÈRES ET AL.**

(iii) simultaneously adding 8-ethoxycarbonyloctanol and the base. After *in situ* desilylation, this procedure gave the target  $\alpha$ -mannopyranoside  $\delta \alpha$  in 55% overall yield from D-Man, i.e., approximately 90% for each chemical step. Finally, selective 4,6-*O*-benzylidenation was achieved by transacetalation using benzaldehyde dimethyl acetal and affording selectively acceptor **3**.

Having the required building blocks in hand, we next investigated the coupling reaction. Furanosylation of 8-ethoxycarbonyloctanol, yielding the protected galactofuranoside **9**, was first achieved under standard activation conditions, i.e., *N*-iodosuccinimide (NIS) and trimethylsilyl trifluoromethanesulfonate (TMSOTf), of both phenyl and ethyl thiogalactofuranosides **2a** and **2b**, respectively (Scheme 4). A better glycosylation yield was obtained from ethyl derivative **2b** (52% vs 40%), thus denoting a higher reactivity of **2b** over **2a**. <sup>12</sup> In this context, glycosylation of diol acceptor **3** was performed using **2b** as the furanosyl donor. Bis-glycofuranosylation of the mannosidic acceptor was avoided by carrying out the coupling reaction at  $0^{\circ}$ C. Under these conditions, the desired  $\beta$ - $(1\rightarrow 3)$ -disaccharide **10** was obtained with high regioselectivity (no trisaccharide was identified) and isolated in 46 % yield.

Finally, deprotection of hydroxyl groups and transesterification of the ethyl ester of compound **9** were simultaneously achieved under Zemplen conditions to afford the desired galactofuranoside **11**. Moreover, a similar debenzoylation procedure was followed by hydrogenolytic removal of the 4,6-benzylidene acetal catalyzed over Pd/C that allowed the synthesis of the targeted hapten **1** in 90% yield for the last two steps.

Compounds **9**, **10**, **11** and **1** were easily characterized on the grounds of signals identified by COSY and heteronuclear  ${}^{1}H^{-13}C$  2D experiments. The  $\beta$  config-



**Reagents and conditions:** (a)  $HO(CH_2)_2CO_2Et$ , NIS, TMSOTf, 4Å MS,  $CH_2Cl_2$  (40%) from **2a**, 62% from **2b**); (b) **3**, NIS, TMSOTf,  $4\text{\AA}$  MS,  $CH_2Cl_2$  (46%); (c) NaOMe, MeOH (93%); (d) NaOMe, MeOH; H<sub>2</sub> Pd/C (90%).



MARCEL DEKKER, INC.

270 Madison Avenue, New York, New York 10016



uration of the furanosyl entities was based on a small coupling constant between H-1' and H-2'  $(J_{1',2'} \approx 0$  Hz). Moreover, glycosylation of the equatorial hydroxyl of **3** was evidenced, for disaccharide **10**, by a typical downfield chemical shift for C-3 ( $\delta_{C-3}$  = 72.6 ppm for **3** vs.  $\delta_{C-3}$  = 76.8 ppm for **10**) as well as upfield shift for both adjacent carbon atoms C-2 ( $\delta$ <sub>C-2</sub> = 72.2 ppm for **3** vs.  $\delta$ <sub>C-2</sub> = 68.7 ppm for **10**) and C-4 ( $\delta_{C-4}$  = 68.5 ppm for **3** vs.  $\delta_{C-4}$  = 72.0 ppm for **10**).<sup>(13)</sup> A similar behaviour was observed by comparison of C-3, C-2 and C-4 chemical shifts from compounds **8** and **1**, respectively.

In summary, we have developed an efficient route toward the synthesis of a glycosidic hapten characterized by the disaccharide  $\beta$ -D-Galf-(1→3)- $\alpha$ -D-Manp which mimics the non-reducing and epitopic part of the main lipopeptidophosphoglycan (LPPG) found in *T. cruzi*. Particular attention has been given to establishing (i) the specific formation of the furanoid ring and (ii) a short synthesis of the mannopyranosyl moiety. The key step of our strategy consisted in a four step (activation, anomerization, diastereospecific glycosidation without anchimeric assistance of participating groups, and deprotection)–one pot preparation of mannoside **8** and resulted in an important decrease in the number of purification and isolation steps required and in a significantly increased overall yield. Continuing efforts are currently under way for the chemical synthesis of natural derivatives containing hexofuranosyl residues.

#### **EXPERIMENTAL**

**General Methods.** While all chemicals were commercially available and used as received, octyl D-galactofuranoside (**5**) was prepared according to ref. 6. All reactions were performed under a nitrogen atmosphere. TLC analyses were carried out on precoated non-activated plates (E. Merck 60  $F_{254}$ ) with detection by UV absorption (254 nm), when applicable, and charring with 5% sulfuric acid in ethanol. For column chromatography, E. Merck 60H  $(5-40 \mu m)$  silica gel was used. Optical rotations were determined with a Perkin-Elmer 341 polarimeter at 20 °C using a 1 dm cell. Melting points were determined using a Reichert microscope with heating plate and are uncorrected.  ${}^{1}H$  and  ${}^{13}C$  NMR spectra were recorded on a Bruker ARX 400 spectrometer at 400 and 100 MHz, respectively. Chemical shifts are given in ppm  $(\delta)$ . CDCl<sub>3</sub> or CD<sub>3</sub>OD and trimethylsilane were used as solvent and internal standard, respectively. Microanalyses were performed by the Service de Microanalyses de l'ICSN (Gif sur Yvette, France).

*n***-Octyl 2,3,5,6-Tetra-***O***-benzoyl-D-galactofuranoside (6).** To a solution of  $5^6$  (1.53 g, 5.20 mmol, $\beta/\alpha$ =1.86:1) in dry pyridine (17 mL) cooled at 0°C was added benzoyl chloride (2.92 mL, 25.15 mmol). After stirring for 3 h and concentration under reduced pressure, the resulting crude mixture was diluted with dichloromethane. The organic layer was successively washed with 5% aqueous HCl, saturated aqueous  $K_2CO_3$  and water, dried (MgSO<sub>4</sub>), and concentrated. Flash-chromatography (9:1 light petroleum/ethyl acetate) then afforded **6** (2.70



MARCEL DEKKER, INC.

270 Madison Avenue, New York, New York 10016



 $g, \beta/\alpha = 1.9:1$ ) in 73% yield as an amorphous solid. TLC (4:1 light petroleum/ethyl acetate) Rf 0.6.  $6\beta$ : <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.08 – 7.86 (m, 8 H, C<sub>6</sub>H<sub>5</sub>); 7.58 – 7.18 (m, 12 H, C<sub>6</sub>H<sub>5</sub>); 6.10–6.05 (m, 1 H, H-5); 5.62 (d, 1 H, H-3, *J*<sub>3,4</sub>=5.1 Hz); 5.46 (s, 1 H, H-1); 5.30 (s, 1 H, H-2); 4.78 (dd, 1 H, H-6a,  $J_{6a,6b}$ =11.7 Hz,  $J_{6a,5}$ =4.6 Hz); 4.73 (dd, 1 H, H-6b,  $J_{6b,5}$ =6.6 Hz); 4.63 (dd, 1 H, H-4,  $J_{4,5}$ =3.6 Hz); 3.73 (dt, 1 H, OCH<sub>2</sub>CH<sub>2</sub>, <sup>2</sup>J=9.2 Hz, <sup>3</sup>J=7.1 Hz); 3.53 (dt, 1 H, OCH<sub>2</sub>CH<sub>2</sub>, <sup>3</sup>J=6.1 Hz); 1.67 – 1.54 (m, 2 H, OCH<sub>2</sub>CH<sub>2</sub>); 1.40 – 1.15 [m, 10 H, (CH<sub>2</sub>)<sub>5</sub>]; 0.86 (t, 3 H, CH<sub>3</sub>,  $\frac{3}{4}$  = 6.6 H<sub>7</sub>)<sup>-13</sup>C NMR (CDC<sub>L)</sub>  $\frac{3}{4}$  166 1 165 7 165 5 (CO): 133 4 133 3 133 2  $3J=6.6$  Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  166.1, 165.7, 165.5 (CO); 133.4, 133.3, 133.2, 133.1 (C<sub>ipso</sub>); 130.0 – 128.3 (C<sub>6</sub>H<sub>5</sub>); 105.6 (C-1); 82.1 (C-2); 81.2 (C-4); 77.6 (C-3); 70.3 (C-5); 67.6 (OCH<sub>2</sub>CH<sub>2</sub>); 63.5 (C-6); 31.8, 29.5, 29.4, 29.3, 26.2, 22.7 [(CH<sub>2</sub>)<sub>6</sub>]; 14.1 (CH<sub>3</sub>). **6**α: <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 166.0, 165.9, 165.7, 165.5 (CO); 133.4 – 128.2 (C<sub>6</sub>H<sub>5</sub>); 100.5 (C-1); 78.5, 77.8 (C-2, C-4); 74.9 (C-3); 71.9, 69.1 (C-5, OCH<sub>2</sub>CH<sub>2</sub>); 63.0 (C-6); 31.8, 29.3, 29.2, 29.1, 25.9, 22.6 [(CH<sub>2</sub>)<sub>6</sub>]; 14.0 (CH<sub>3</sub>). Anal. Calcd for  $C_{42}H_{46}O_{10}$  (710.83): C, 70.97; H, 6.52. Found: C, 70.65; H,

6.50.

**2,3,5,6-Tetra-***O***-benzoyl-1-***O***-trifluoroacetyl-D-galactofuranose (7).** To a solution of **6** (1.585 g, 2.24 mmol) in dry dichloromethane (24 mL) were added, at 0°C, trifluoroacetic anhydride (1.27 mL, 8.95 mmol) and sulfuric acid (358  $\mu$ L, 6.72 mmol). After stirring for 35 min, neutralization with triethylamine and concentration under reduced pressure, the residue was subjected to column chromatography (5:5:0.01 light petroleum/ethyl acetate/triethylamine) that afforded an anomeric mixture of  $7(1.24 \text{ g}, 80\%, \beta/\alpha=4.6:1)$  as an amorphous compound. This product could be characterized by NMR analysis and thus used for subsequent reaction without further purification. TLC (5:5:0.01 light petroleum/ethyl acetate/triethylamine) Rf 0.4. **7** $\beta$ : <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.10–7.26 (m, 20 H, C<sub>6</sub>H<sub>5</sub>); 6.07 (dt, 1 H, H-5,  $J_{4,5}$ =4.0 Hz,  $3J$ =6.6 Hz); 5.71 (s, 1 H, H-1); 5.64 (d, 1 H, H-3,  $J_{3,4}$ =5.1 Hz); 5.51 (s, 1 H, H-2); 4.86 (dd, 1 H, H-4); 4.78 (dd, 1 H, H-6a,  $J_{5.6a}$ =4.6 Hz,  $J_{6a,6b}$ =12.0 Hz); 4.71 (dd, 1 H, H-6b,  $J_{5,6b}$ =7.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  166.2, 165.8, 165.7, 165.6 (CO); 133.7-128.3 (C<sub>6</sub>H<sub>5</sub>); 100.9 (C-1); 82.7, 81.6 (C-4, C-2); 77.7 (C-3); 70.5 (C-5); 63.6 (C-6). <sup>19</sup>F NMR (CDCl3)  $\delta$  –75.5 (CF<sub>3</sub>). **7** $\alpha$ : <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 8.10−7.26 (m, 20 H, C<sub>6</sub>H<sub>5</sub>); 6.12 (t, 1 H, H-2, *J*<sub>1,2</sub>=*J*<sub>2,3</sub>=4.6 Hz); 5.97 (1 H, H-5,  $J_{4,5}$ =6.4 Hz,  $3J$ =3.6 Hz); 5.84 (d, 1 H, H-1); 5.54 (t, 1 H, H-3, *J*<sub>3,4</sub>=4.6 Hz); 4.84 (dd, 1 H, H-6a, *J*<sub>6a,6b</sub>=12.2 Hz); 4.72 (dd, 1 H, H-6b); 4.56 (dd, 1 H, H-4). 13C NMR (CDCl3) 166.3, 166.0, 165.7, 165.5 (CO); 95.9 (C-1); 79.2, 77.6 (C-4, C-2); 75.7 (C-3); 72.2 (C-5); 63.1 (C-6). <sup>19</sup>F NMR (CDCl<sub>3</sub>)  $\delta$  -75.5  $(CF_3)$ .

**Phenyl 2,3,5,6-Tetra-O-benzoyl-1-thio-β-D-galactofuranoside (2a).** To a solution of **7** (1.80 g, 2.60 mmol) in dry dichloromethane (18 mL) were successively added, at  $0^{\circ}$ C, thiophenol (0.40 mL, 3.9 mmol) and  $BF_3$ . OEt<sub>2</sub> (0.66 mL, 5.2 mmol). The reaction was then monitored by TLC (3:2 light petroleum/ethyl acetate). After complete consumption of **7**, the mixture was diluted with dichloromethane (75 mL), washed with 5% aqueous NaHCO<sub>3</sub> and with water. The organic layer was finally dried  $(MgSO<sub>4</sub>)$ , concentrated, and the residue was chro-





Downloaded At: 07:09 23 January 2011 Downloaded At: 07:09 23 January 2011

matographically (4:1 light petroleum/ethyl acetate) purified. This procedure gave 2.00 g (89%) of desired **2a**. TLC (3:2 light petroleum/ethyl acetate) R*f* 0.7. mp =98–100 °C (ethyl acetate/light petroleum).  $[\alpha]_D^{20}$  –66 (*c* 1.0, CH<sub>2</sub>Cl<sub>2</sub>). <sup>1</sup>H NMR  $(CDCl_3)$   $\delta$  8.06–7.21 (m, 25 H, C<sub>6</sub>H<sub>5</sub>); 6.11 (dt, 1 H, H-5,  $J_{5,6b}$ =6.9 Hz, *J*<sub>5,6a</sub>=*J*<sub>5,4</sub>=4.6 Hz); 5.84 (s, 1 H, H-1); 5.71 (d, 1 H, H-3, *J*<sub>3,4</sub>=5.0 Hz); 5.67 (s, 1 H, H-2); 4.95 (t, 1 H, H-4); 4.76 (dd, 1 H, H-6a,  $J_{6a,6b}$ =11.8 Hz); 4.71 (dd, 1 H, H-6b). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  166.0, 165.7, 165.5, 165.3 (CO); 133.6, 133.5, 133.3, 133.1 (COC<sub>ipso</sub>); 132.4 (SC<sub>ipso</sub>); 130.0–127.3 (C<sub>6</sub>H<sub>5</sub>); 91.3 (C-1); 82.4 (C-2); 81.5 (C-4); 77.8 (C-3); 70.3 (C-5); 63.4 (C-6).

Anal. Calcd for  $C_{40}H_{32}O_9S$  (688.76): C, 69.76; H, 4.68. Found C, 69.83; H, 4.69.

**Ethyl** 2,3,5,6-Tetra-*O*-benzoyl-1-thio-β-D-galactofuranoside (2b). Product **2b** was prepared according to the previous procedure starting with **7** (1.80 g, 2.60 mmol), ethanethiol (0.45 mL, 5.9 mmol) and  $BF_3$ . OEt<sub>2</sub> (0.66 mL, 5.2) mmol). Work-up and chromatographic (4:1 light petroleum/ethyl acetate) purification afforded **2b** (1.20 g) in 72% yield. TLC (7:3 light petroleum/ethyl acetate)  $Rf 0.7. [\alpha]_D^{20} - 26$  (*c* 1.0,  $CH_2Cl_2$ ). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.10 – 7.26 (m, 20 H, C<sub>6</sub>H<sub>5</sub>); 6.13–6.09 (m, 1 H, H-5); 5.67 (d, 1 H, H-1,  $J_{1,2}$ =1.0 Hz); 5.67–5.66 (m, 1 H, H-3); 5.50 (t, 1 H, H-2,  $J_{2,3}=1.0$  Hz); 4.84 (t, 1 H, H-4,  $J_{3,4}=J_{4,5}=4.0$  Hz); 4.77 (dd, 1 H, H-6a,  $J_{6a,6b}$ =11.7 Hz,  $J_{6a,5}$ =4.6 Hz); 4.74 (dd, 1 H, H-6b,  $J_{6b,5}$ =6.6 Hz); 2.78 (dq, 1 H, CH<sub>2</sub>CH<sub>3</sub>, <sup>2</sup>J=12.7 Hz, <sup>3</sup>J=7.6 Hz); 2.69 (dq, 1 H, CH<sub>2</sub>CH<sub>3</sub>, <sup>3</sup>J=7.6 Hz); 1.33 (t, 3 H, CH<sub>2</sub>CH<sub>3</sub>). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 166.1, 165.7, 165.5, 165.4 (CO); 133.4, 133.3, 133.2, 133.1 (Cipso); 130.1, 130.0, 129.9, 129.7, 128.5, 128.4, 128.3  $(C_6H_5)$ ; 88.2 (C-1); 82.9 (C-2); 81.1 (C-4); 77.9 (C-3); 70.2 (C-5); 63.5 (C-6); 25.3  $(CH_2CH_3)$ ; 14.9  $(CH_2CH_3)$ .

Anal Calcd for C<sub>36</sub>H<sub>32</sub>O<sub>9</sub>S (640.71): C, 67.49; H 5.03. Found C, 67.49; H, 5.19.

**8-Ethoxycarbonyloctyl**  $\alpha$ -D-Mannopyranoside (8). To a solution of Dmannose (5.00 g, 27.8 mmol) in dry pyridine (80 mL) were successively added hexamethyldisilazane (32.2 mL, 152.8 mmol) and chlorotrimethylsilane (19.4 mL, 152.8 mmol). After stirring at room temperature for 35 min, the solvent was removed under reduced pressure. The resulting mixture was then diluted with  $Et_2O$ (100 mL) and washed with 5 % aqueous HCl, saturated aqueous NaHCO<sub>3</sub> and water. The organic layer was finally dried  $(MgSO<sub>4</sub>)$  and concentrated to give an anomeric mixture of trimethylsilyl 2,3,4,6-tetra-*O*-trimethylsilyl-D-mannopyranoside (6) (14.87 g, $\alpha/\beta$ =6.7:1). TLC (9:1 light petroleum/diethylether) Rf 0.8. Selected <sup>1</sup>H NMR data<sup>14</sup> for 6: (CDCl<sub>3</sub>)  $\delta$  4.90 (s, 0.87 H, H-1 $\alpha$ ); 4.64 (s, 0.13 H, H-1β). Selected <sup>13</sup>C NMR data<sup>14</sup> for **6**: (CDCl<sub>3</sub>) δ 95.6 (C-1 $\alpha$ ); 95.5 (C-1β).

The crude oil (3.93 g, 7.26 mmol) was diluted in dry dichloromethane (20 mL) before adding freshly distilled iodotrimethylsilane (1.1 mL, 7.99 mmol). The mixture was stirred at room temperature for 45 min, tetrabutylammonium iodide (2.68 g, 7.26 mmol) was added and stirred for 15 min more before adding a dichloromethane (20 mL) solution of 8-ethoxycarbonyloctan-1-ol (2.93 g, 14.5





#### **862 FERRIÈRES ET AL.**

mmol) and 1.49 g (7.26 mmol) of 2,6-di-*tert*-butyl-4-methylpyridine. The reaction mixture was stirred at room temperature for 6 h, and final desilylation was performed in 30 min with methanol (60 mL). The resulting solution was made neutral (triethylamine), concentrated, and purification by column chromatography (9:1 dichloromethane/methanol) afforded the target compound **8** (1.45g) in 55% global yield. TLC (8:1 dichloromethane/methanol) Rf 0.3.  $[\alpha]_D^{20} + 28$  (*c* 1.0, methanol) [lit.<sup>15</sup> [ $\alpha$ ]<sup>25</sup> +48 (*c* 0.75, water)]. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.70 (s, 1 H, H-1); 4.10–4.05 (m, 2 H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); 3.79–3.72 (m, 2 H, H-2, H-6a); 3.70–3.62 (m, 3 H, H-3, H-6b, OCH<sub>2</sub>CH<sub>2</sub>); 3.58 (t, 1 H, H-4,  $J_{3,4} = J_{4,5} = 9.5$  Hz); 3.50 – 3.43 (m, 1 H, H-5); 3.40–3.35 (m, 1 H, OCH<sub>2</sub>CH<sub>2</sub>); 2.27 (t, 2 H, CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Et, <sup>3</sup>J=7.4 Hz); 1.62 - 1.50 [m, 4 H,  $(CH_2)_2$ ]; 1.38 - 1.24 [m, 8 H,  $(CH_2)_4$ ]; 1.21 (t, 3 H,  $CO_2CH_2CH_3$ ,  ${}^{3}J=7.1$  Hz). <sup>13</sup>C NMR (CD<sub>3</sub>OD)  $\delta$  175.5 (CO); 101.4 (C-1); 74.4 (C-5); 72.6 (C-3); 72.2 (C-2); 68.5 (C-4, OCH<sub>2</sub>CH<sub>2</sub>); 62.8 (C-6); 61.3 (CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); 35.0 (CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Et); 30.5, 30.3, 30.2, 30.0, 27.2, 26.0 [(CH<sub>2</sub>)<sub>6</sub>]; 14.6 ( $CO_2CH_2CH_3$ ).

**8-Ethoxycarbonyloctyl 4,6-***O***-Benzylidene--D-mannopyranoside (3).** To a solution of **8** (0.90 g, 2.47 mmol) in ethyl acetate (5 mL) were successively added  $ZnCl_2$  (0.36 g, 2.64 mmol) and benzaldehyde dimethyl acetal (408  $\mu$ L, 2.64 mmol). After stirring at 50 °C for 1.5 h, the reaction mixture was cooled and then washed with 10% aqueous  $Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>$ , saturated aqueous NaHCO<sub>3</sub> and water. The organic layer was dried  $(MgSO<sub>4</sub>)$  and concentrated under reduced pressure. The target compound **3** was finally purified by column chromatography (3:2 light petroleum/ethyl acetate) and isolated in 44% yield (488 mg) as a colorless oil. TLC (1:1 light petroleum/ethyl acetate) Rf 0.4.  $[\alpha]_D^{20} + 37$  (*c* 1.2, CHCl<sub>3</sub>) [lit.<sup>9</sup>  $[\alpha]_D^{20}$  $+37$  (*c* 1.2, CHCl<sub>3</sub>)]. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  7.51 – 7.48 (m, 2 H, C<sub>6</sub>H<sub>5</sub>); 7.39 – 7.35 (m, 3 H, C<sub>6</sub>H<sub>5</sub>); 5.56 (s, 1 H, PhC*H*); 4.82 (s, 1 H, H-1); 4.29 – 4.23 (m, 1 H, H-6a); 4.14−4.09 (m, 2 H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); 4.10−4.04 (m, 1 H, H-3); 4.01−3.98 (m, 1 H, H-2); 3.91 (t, 1 H, H-4,  $J_{3,4} = J_{4,5} = 9.1$  Hz); 3.83 – 3.79 (m, 2 H, H-6b, H-5); 3.68  $(\text{dt}, 1 \text{ H}, \text{OCH}_2\text{CH}_2, {}^2J=9.6 \text{ Hz}, {}^3J=6.6 \text{ Hz})$ ; 3.40 (dt, 1 H, OC*H*<sub>2</sub>CH<sub>2</sub>, <sup>3</sup>J=6.6 Hz); 2.29 (t, 2 H, CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Et, <sup>3</sup>J=7.6 Hz); 1.64 – 1.56 [m, 4 H, (CH<sub>2</sub>)<sub>2</sub>]; 1.38 – 1.29  $[m, 8 H, (CH<sub>2</sub>)<sub>4</sub>]; 1.25$  (t, 3 H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, <sup>3</sup>J=7.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ 174.0 (CO); 137.2 (Cipso); 129.2, 128.3, 126.2 (C6H5); 102.2 (Ph*C*H); 100.1 (C-1); 77.0 (C-4); 71.1 (C-3); 68.8 (C-6); 68.6 (C-3); 68.0 (OCH<sub>2</sub>CH<sub>2</sub>); 63.0 (C-5); 60.2 (CO2*C*H2CH3); 34.3 (CH2*C*H2CO2Et); 29.3, 29.1, 29.0, 26.0, 24.9 [(CH2)6]; 14.2  $(CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>).$ 

**8-Ethoxycarbonyloctyl 2,3,5,6-Tetra-***O***-benzoyl-**-**-D-galactofuranoside (9).** To a solution of donor **2b** (0.093 g, 0.145 mmol) and 8-ethoxycarbonyloctanol (0.024 g, 0.212 mmol) in dry dichloromethane (1 mL) in the presence of molecular sieves (4 Å, 0.20 g), cooled at  $0^{\circ}$ C and protected from light, were successively added *N*-iodosuccinimide (26 mg, 0.145 mmol) and trimethylsilyl triflate (2  $\mu$ L, 0.012 mmol). After stirring for 3 h at 0°C and 4 h at room temperature, the reaction mixture was made neutral by a few drops of triethylamine, filtered over a bed of celite, concentrated and finally purified by column chromatography (7:2:1





Downloaded At: 07:09 23 January 2011 Downloaded At: 07:09 23 January 2011

light petroleum/ethyl acetate/dichloromethane). Product **9** (0.059 g) was thus isolated in 52% yield. TLC (7:2:1 light petroleum/ethyl acetate/dichloromethane) R*f* 0.4. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.09 – 7.89 (m, 8 H, C<sub>6</sub>H<sub>5</sub>); 7.59 – 7.26 (m, 12 H, C<sub>6</sub>H<sub>5</sub>); 6.10–6.06 (m, 1 H, H-5); 5.63 (d, 1 H, H-3, *J*<sub>3,4</sub>=5.1 Hz); 5.47 (s, 1 H, H-1); 5.30 (s, 1 H, H-2); 4.79 (dd, 1 H, H-6a,  $J_{6a,6b}$  = 11.7 Hz,  $J_{6a,5}$  = 4.6 Hz); 4.74 (dd, 1 H, H-6b,  $J_{6b,5}$ =7.1 Hz); 4.64 (dd, 1 H, H-4,  $J_{4,5}$ =3.6 Hz); 4.15-4.09 (m, 2 H,  $CO_2CH_2CH_3$ ); 3.75 (dt, 1 H,  $OCH_2CH_2$ ,  $^2J=9.6$  Hz,  $^3J=6.6$  Hz); 3.53 (dt, 1 H, OCH<sub>2</sub>CH<sub>2</sub>, <sup>3</sup>J=6.1 Hz); 2.26 (t, 2 H, CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Et, <sup>3</sup>J=7.4 Hz); 1.67-1.56 [m, 4 H, (CH<sub>2</sub>)<sub>2</sub>]; 1.41 – 1.21 [m, 8 H, (CH<sub>2</sub>)<sub>4</sub>]; 1.25 (t, 3 H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, <sup>3</sup>J=7,1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 173.8 (*C*OOEt); 166.1, 165.7, 165.6, 165.4 (O*C*OPh); 133.5, 133.3, 133.2, 133.1 (C<sub>ipso</sub>); 129.9–128.3 (C<sub>6</sub>H<sub>5</sub>); 105.5 (C-1); 82.0 (C-2); 81.2 (C-4); 77.6 (C-3); 70.3 (C-5); 67.6 (OCH<sub>2</sub>CH<sub>2</sub>); 63.5 (C-6); 60.1 (CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); 34.3 (CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Et); 29.4, 29.3, 29.2, 29.1, 26.1, 24.9 [(CH<sub>2</sub>)<sub>6</sub>]; 14.2 (CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>). Anal Calcd for C<sub>45</sub>H<sub>48</sub>O<sub>12</sub> (780.87): C, 69.22; H, 6.20. Found C, 69.15; H, 6.25.

**8-Ethoxycarbonyloctyl 2,3,5,6-Tetra-***O***-benzoyl-**-**-D-galactofuranosyl-**  $(1(3)-4,6-O$ **-benzylidene-** $\alpha$ **-D-mannopyranoside** (10). Galactofuranosyl donor **2b** (0.17 g, 0.27 mmol) and glycosyl acceptor **3** (0.10 g, 0.22 mmol) were dissolved in anhydrous dichloromethane (4 mL) containing molecular sieves (4 Å, 0.20 g). The reaction mixture was then protected from light and cooled to  $0^{\circ}$ C before adding *N*-iodosuccinimide (60 mg, 0.27 mmol) and trimethylsilyl triflate (9  $\mu$ L, 0.04 mmol) and stirring for 15 min. After completion of the reaction, the media was quenched with several drops of triethylamine until it turned into a yellow solution. The resulting mixture was filtered over a bed of celite, concentrated under reduced pressure and finally subjected to column chromatography (4:1 toluene/ethyl acetate). This procedure provided the required disaccharide **10** (0.103 g) in 46% yield. TLC (4:1 toluene/ethyl acetate)  $Rf(0.3. [\alpha]_D^{20} + 61$  (*c* 1.0, CHCl<sub>3</sub>). <sup>1</sup>H NMR  $(CDCl_3)$   $\delta$  8.04 – 7.81 (m, 8 H, C<sub>6</sub>H<sub>5</sub>); 7.59 – 7.16 (m, 17 H, C<sub>6</sub>H<sub>5</sub>); 5.90 (ddd, 1 H, H-5',  $J_{5,6,6}$ <sup>-3</sup> = 8.6 Hz,  $J_{5,6,6}$  = 5.6 Hz,  $J_{5,4}$  = 3.0 Hz); 5.56 (dd, 1 H, H-3',  $J_{3,4,5}$  = 5.6 Hz,  $J_{3',2'}=1.5$  Hz); 5.49 (s, 1 H, H-1'); 5.46 (s, 1 H, PhC*H*); 5.44 (d, 1 H, H-2'); 4.93 (s, 1 H, H-1); 4.74 (dd, 1 H, H-4'); 4.59 (dd, 1 H, H-6'a,  $J_{6' a, 6' b} = 12.2$  Hz,); 4.31 – 4.26 (m, 2 H, H-4, H-5); 4.16 – 4.02 (m, 5 H, H-2, H-3, H-6'b, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); 3.92 (dd, 1 H, H-6a,  $J_{6a,6b}$ =10.2 Hz,  $J_{6a,5}$ =4.6 Hz); 3.90 (dd, 1H, H-6b,  $J_{6b,5}$ =5.1 Hz); 3.71 (dt, 1 H, OC*H*<sub>2</sub>CH<sub>2</sub>, <sup>2</sup>J=9.6 Hz, <sup>3</sup>J=6.6 Hz); 3.41 (dt, 1 H, OCH<sub>2</sub>CH<sub>2</sub>, <sup>3</sup>J=6.6 Hz); 2.28 (t, 2 H, CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Et, <sup>3</sup>J=7.1 Hz); 1.62-1.58 [m, 4 H, (CH<sub>2</sub>)<sub>2</sub>]; 1.36 – 1.21 [m, 8 H, (CH<sub>2</sub>)<sub>4</sub>]; 1.23 (t, 3 H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, <sup>3</sup>J=7.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 166.0, 165.9, 165.7, 165.5 (CO); 137.3 (CHC<sub>ipso</sub>); 135.5, 133.5, 133.1, 132.3 (CO*C*ipso); 130.0 125.9 (C6H5); 102.4 (C-1); 102.0 (Ph*C*H); 100.1 (C-1); 82.4 (C-2); 81.3 (C-4); 77.2 (C-3); 76.8 (C-3); 72.0 (C-4); 70.1 (C-5'); 68.9 (C-5); 68.7 (C-2); 68.1 (OCH<sub>2</sub>CH<sub>2</sub>); 64.2 (C-6'); 63.6 (C-6); 60.1 (CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); 34.3 (CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Et); 30.3, 29.7, 29.2, 29.1, 26.0, 24.9 [(CH<sub>2</sub>)<sub>6</sub>]; 14.2 ( $CO_2CH_2CH_3$ ).

Anal. Calcd for  $C_{58}H_{62}O_{16}$  (1015.13): C, 68.62; H, 6.16. Found C, 68.45; H, 6.20.





**8-Methoxycarbonyloctyl β-D-Galactofuranoside (11).** To a solution of **9**  $(0.050 \text{ g}, 64 \mu \text{mol})$  in anhydrous methanol  $(2 \text{ mL})$  was added sodium  $(0.1 \text{ mg})$ . After stirring at room temperature for 24 h, the reaction was made neutral with IR120- H-form resin, filtered and concentrated under reduced pressure. After removal of the solvent, the desired product **11** was purified by flash-chromatography (9:1 dichloromethane/methanol) and isolated (0.021 g) in 93% yield. TLC (9:1 dichloromethane/methanol) Rf 0.3.  $[\alpha]_D^{20}$  – 66 (*c* 1.0, MeOH) [lit<sup>9</sup> [ $\alpha]_D^{20}$  – 73 (*c* 0.6, EtOH)]. <sup>1</sup>H NMR (CD<sub>3</sub>OD) δ 4.75 (d, 1 H, H-1, *J*<sub>1,2</sub>=1.8 Hz); 3.90 (dd, 1 H, H-3, *J*<sub>2,3</sub>=4.0 Hz, *J*<sub>3,4</sub>=6.6 Hz); 3.83 (dd, 1 H, H-2); 3.81 (dd, 1 H, H-4, *J*<sub>4,5</sub>=3.0 Hz);  $3.65 - 3.50$  (m, 7 H, H-5, H-6, H-6', H- $\alpha$ , CH<sub>3</sub>);  $3.37 - 3.28$  (m, 1 H, H- $\alpha'$ ); 2.22 (t, 2 H, CH<sub>2</sub>CO, <sup>3</sup>J=7.5 Hz); 1.52-1.47 (m, 2 H, CH<sub>2</sub> $\beta$ ); 1.25-1.21 [m, 8 H,  $(CH<sub>2</sub>)<sub>4</sub>$ ]. <sup>13</sup>C NMR (CD<sub>3</sub>OD)  $\delta$  176.0 (CO); 109.4 (C-1); 84.0 (C-4); 83.4 (C-2); 78.7 (C-3); 72.4 (C-5); 68.9 (CH<sub>2</sub> $\alpha$ ); 64.6 (C-6); 52.0 (CH<sub>3</sub>); 34.8 (CH<sub>2</sub> $\beta$ ); 30.7,  $30.3, 30.1, 27.2, 26.0$  [ $(CH<sub>2</sub>)<sub>6</sub>$ ].

**8-Methoxycarbonyloctyl β-D-Galactofuranosyl-(1→3)-α-D-mannopyranoside (1).** To a solution of disaccharide **10** (0.10 g, 0.097 mmol) in anhydrous methanol (2 mL) was added a 0.1 M methanolic solution of sodium methoxide (0.97 mL). After stirring at room temperature for 16 h, the reaction mixture was made neutral with  $IR120-H^+$ -form resin, filtered and concentrated under reduced pressure: TLC (9:1 dichloromethane/methanol) R*f* 0.4. The resulting crude product was then subjected to hydrogenolysis in absolute ethanol  $(2 \text{ mL})$  using  $10\%$ palladium activated on charcoal as catalyst. No transesterification was observed under these conditions. After 48 h at room temperature, the catalyst was removed by filtration, and the solvent was evaporated. A chromatographic purification (4:1 dichloromethane/methanol) provided **1** (45 mg, 90% over two steps): TLC (4:1 CH<sub>2</sub>Cl<sub>2</sub>/MeOH) Rf 0.4. [ $\alpha$ ]<sub>120</sub><sup>2</sup> + 61 (*c* 1.0, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CD<sub>3</sub>OD)  $\delta$  4.99 (s, 1 H, H-1'); 4.69 (d, 1 H, H-1,  $J_{1,2} = 1.5$  Hz); 4.02 – 3.98 (m, 1 H, H-3'); 3.97 – 3.94 (m, 2 H, H-2', H-4'); 3.89 - 3.87 (m, 1 H, H-2); 3.79 - 3.73 (m, 2 H, H-3, H-6a); 3.67–3.46 (m, 7 H, H-5', H-6'a, H-4, H-5, OC*H*<sub>2</sub>CH<sub>2</sub>); 3.57 (s, 3 H, CO<sub>2</sub>CH<sub>3</sub>); 3.35 (dt, 1 H, OCH<sub>2</sub>CH<sub>2</sub>, <sup>2</sup>J=9.6 Hz, <sup>3</sup>J=6.6 Hz); 2.24 (t, 2 H, CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Me, <sup>3</sup>J=7.1 Hz); 1.57 – 1.49 [m, 4 H, (CH<sub>2</sub>)<sub>2</sub>]; 1.34 – 1.22 [m, 8 H, (CH<sub>2</sub>)<sub>4</sub>]. <sup>13</sup>C NMR (CD<sub>3</sub>OD) δ 176.1 (CO); 106.5 (C-1'); 101.4 (C-1); 85.4 (C-4'); 82.8 (C-2'); 79.0 (C-3'); 77.5 (C-3); 74.5 (C-5); 72.5 (C-5'); 68.8 (C-2); 68.7 (OCH<sub>2</sub>CH<sub>2</sub>); 66.8 (C-4); 64.4 (C-6'); 62.9 (C-6); 52.0 (CO<sub>2</sub>CH<sub>3</sub>); 34.8 (CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>Me); 30.6, 30.4, 30.3, 30.1, 27.3, 26.0 [(CH<sub>2</sub>)<sub>6</sub>]. [ $\alpha$ ]<sup>20</sup> (of peracetylated disaccharide) +12 (*c* 1.0, CHCl<sub>3</sub>) [lit.<sup>9</sup> [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 37 (*c* 1.2, CHCl<sub>3</sub>)].

#### **ACKNOWLEDGMENTS**

The authors are grateful to Thierry Benvegnu for helpful discussion, to Martine Lefeuvre for assistance in NMR analysis and to Jula Meyer (Erasmus fellowship student) for her contribution to this project.

MARCEL DEKKER, INC.

270 Madison Avenue, New York, New York 10016



Downloaded At: 07:09 23 January 2011 Downloaded At: 07:09 23 January 2011

#### **REFERENCES**

- 1. Dedicated to Gérard Descotes for his contribution to Carbohydrate Chemistry.
- 2. Ferrières, V.; Gelin, M.; Roussel, M.; Plusquellec, D. Abstracts of Papers, Second EuroConference on Carbohydrates in Drug Research, Estoril, Portugal, Sept. 14–17, 2000; C16.
- 3. Guiguen, C. *Parasitologie–Mycologie Maladies parasitaires et fongiques*, 5th Ed.; Association Française des Professeurs de Parasitologie et C. & R., Eds; La Madeleine, 1992; 481 pp.
- 4. de Lederkremer, R.M.; Colli, W. Galactofuranose-containing glycoconjugates in trypanosomatids. Glycobiology **1995**, *5*(6), 547–552.
- 5. (a) Gelin, M.; Ferrières, V.; Plusquellec, D. Synthesis of new glycofuranosyl donors and their use in glycosylation reactions. Carbohydr. Lett. **1997**, *2*, 381–388; (b) Velty, R.; Benvegnu, T.; Gelin, M.; Privat, E.; Plusquellec, D. A new convenient synthesis of disaccharides containing furanoside units. Carbohydr. Res. **1997**, *299*, 7–14; (c) Gelin, M.; Ferrières, V.; Plusquellec, D. A general and diastereoselective synthesis of 1,2-*cis*-hexofuranosides from 1,2-*trans*thiofuranosyl donors. Eur. J. Org. Chem. **2000**, 1423–1431.
- 6. Ferrières, V.; Bertho; J. N.; Plusquellec, D. A convenient synthesis of alkyl D-glycofuranosiduronic acids and alkyl D-glycofuranosides from unprotected carbohydrates. Carbohydr. Res. **1998**, *311*, 25–35.
- 7. Ferrières, V.; Gelin, M.; Boulch, R.; Toupet, L.; Plusquellec, D. An efficient route to per-*O*-acetylated hexofuranoses. Carbohydr. Res. **1998**, *314*, 79–83.
- 8. Uchiyama, T.; Hindsgaul, O. Per- $O$ -trimethylsilyl- $\alpha$ -L-fucopyranosyl iodide: A novel glycosylating agent for terminal  $\alpha$ -L-fucosylation. Synlett **1996**, 499–501.
- 9. Tsui, D.S.; Gorin, P.A.J. Preparation of 8-methoxycarbonyloctyl glycosides of  $\alpha$ -Dmannopyranose, 2-O- $\alpha$ -D-mannopyranosyl- $\alpha$ -D-mannopyranose,  $\beta$ -D-galactofuranose, and 3-O-β-D-galactopyranosyl-α-D-mannopyranose. Carbohydr. Res. 1986, 156, 1–8.
- 10. Ikemoto, N.; Kim, O.K.; Lo, L.C.; Satyanarayana, V.; Chang, M.; Nakanishi, K. Ferric chloride, an anomerization catalyst for the preparation of alkyl  $\alpha$ -glycopyranosides. Tetrahedron Lett. **1992**, *33*(30), 4295–42498.
- 11. (a) Lemieux, R.U.; Hayami, J.I. The mechanism of the anomerization of the tetra-*O*acetyl-D-glucopyranosyl chlorides. Can. J. Chem. **1965**, *43*(8), 2162–2173; (b) Lemieux, R.U.; Hendriks, K.B.; Stick, R.V.; James, K. Halide ion catalyzed glycosidation reactions. Syntheses of  $\alpha$ -linked disaccharides. J. Am. Chem. Soc. 1975, *97*(14), 4056–4062.
- 12. Boons, G.J. Strategies in oligosaccharide synthesis. Tetrahedron **1996**, *52* (4), 1095–1121.
- 13. Bock, K.; Pedersen, C. Carbon-13 nuclear magnetic resonance spectroscopy of monosaccharides. Adv. Carbohydr. Chem. Biochem. **1983**, *41*, 27–66.
- 14. Meldal, M.; Knak, M.; Bock, K. Large-scale synthesis of  $\alpha$ -D-mannose 6-phosphate and other hexose 6-phosphates. Carbohydr. Res. **1992**, *235*, 115–127.
- 15. Bochkov, A.F.; Betaneli, V.I.; Kochetkov N.K. Sugar orthoesters. 10. Structure of by-products during glycosylation by orthoesters. Izv. Akad. Nauk. SSSR, Ser. Khim. **1974**, (6), 1379–1386.

Received June 5, 2001 Accepted November 7, 2001



# **Request Permission or Order Reprints Instantly!**

Interested in copying and sharing this article? In most cases, U.S. Copyright Law requires that you get permission from the article's rightsholder before using copyrighted content.

All information and materials found in this article, including but not limited to text, trademarks, patents, logos, graphics and images (the "Materials"), are the copyrighted works and other forms of intellectual property of Marcel Dekker, Inc., or its licensors. All rights not expressly granted are reserved.

Get permission to lawfully reproduce and distribute the Materials or order reprints quickly and painlessly. Simply click on the "Request Permission/Reprints Here" link below and follow the instructions. Visit the [U.S. Copyright Office](http://www.copyright.gov/fls/fl102.html) for information on Fair Use limitations of U.S. copyright law. Please refer to The Association of American Publishers' (AAP) website for guidelines on [Fair Use in the Classroom](http://www.publishers.org/conference/copyguide.cfm).

The Materials are for your personal use only and cannot be reformatted, reposted, resold or distributed by electronic means or otherwise without permission from Marcel Dekker, Inc. Marcel Dekker, Inc. grants you the limited right to display the Materials only on your personal computer or personal wireless device, and to copy and download single copies of such Materials provided that any copyright, trademark or other notice appearing on such Materials is also retained by, displayed, copied or downloaded as part of the Materials and is not removed or obscured, and provided you do not edit, modify, alter or enhance the Materials. Please refer to our [Website](http://www.dekker.com/misc/useragreement.jsp)  [User Agreement](http://www.dekker.com/misc/useragreement.jsp) for more details.

# **[Order now!](http://s100.copyright.com/AppDispatchServlet?authorPreorderIndicator=N&pdfSource=Dekker&publication=CAR&title=A+NEW+APPROACH+TO+A+DISACCHARIDIC+HAPTEN+CONTAINING+A+GALACTOFURANOSYL+ENTITY12&offerIDValue=18&volumeNum=20&startPage=855&isn=0732-8303&chapterNum=&publicationDate=12%2F31%2F2001&endPage=865&contentID=10.1081%2FCAR-100108662&issueNum=9&colorPagesNum=0&pdfStampDate=07%2F28%2F2003+09%3A55%3A12&publisherName=dekker&orderBeanReset=true&author=Vincent+Ferrires%2C+Myriam+Roussel%2C+Muriel+Gelin%2C+Daniel+Plusquellec&mac=JlDd2ghOS8UjX87chXujdg--)**

Reprints of this article can also be ordered at http://www.dekker.com/servlet/product/DOI/101081CAR100108662